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Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
 R factor = 0.052
 wR factor = 0.120
Data-to-parameter ratio = 14.3For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.2-[*N*-(2-Ethoxyphenyl)carbamoylmethoxy]-
N-(2-pyridylmethyl)benzamide

In the title compound, $\text{C}_{23}\text{H}_{23}\text{N}_3\text{O}_4$, the molecule is essentially planar, except for the pyridine ring. In the crystal structure, the molecules are linked into layers by $\text{N}-\text{H}\cdots\text{N}$, $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds. The packing is further stabilized by $\text{C}-\text{H}\cdots\pi$ interactions.

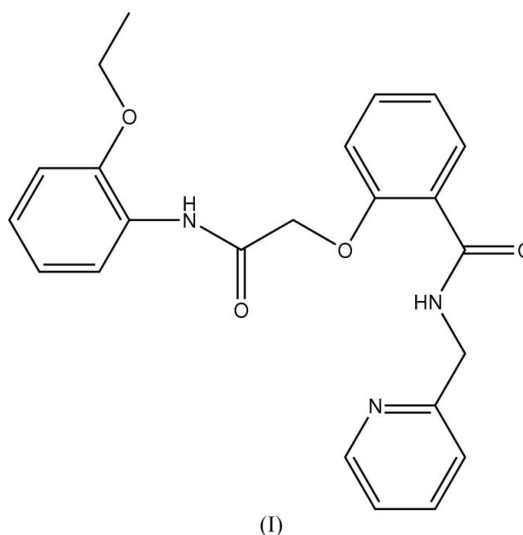
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Comment

We have recently reported the structure of *N,N,N'*-tetraphenyl-2,2'-(*o*-phenylenedioxy)diacetamide, (II) (Wen, Li *et al.*, 2005). Here, we report the structure of the title compound, (I).



Bond lengths and angles in (I) are within normal ranges (Allen *et al.*, 1987) and comparable with those in the related compound, (II). In (I), the non-H atoms, except for those of the pyridine ring, are approximately coplanar, with a dihedral angle of $8.2(1)^\circ$ between the two benzene rings. There are five intramolecular $\text{N}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds (Table 2), contributing to the planarity. The pyridine ring twists from the mean plane, making angles of $71.3(1)$ and $79.4(1)^\circ$, respectively, with the C8–C13 and C16–C21 rings.

In the crystal structure of (I), molecules are linked into layers (Fig. 2) *via* $\text{N}3-\text{H}3\text{B}\cdots\text{N}1^i$, $\text{C}3-\text{H}3\text{A}\cdots\text{O}3^{ii}$ and $\text{C}14-\text{H}14\text{A}\cdots\text{N}1^i$ intermolecular interactions (symmetry codes as in Table 2). The packing is further stabilized by a $\text{C}-\text{H}\cdots\pi$ interaction involving the pyridine ring (centroid Cg1).

Experimental

The title compound was prepared according to the literature method of Wen, Zhang *et al.* (2005). Colourless crystals were obtained by slow evaporation of a petroleum ether–ethyl acetate solution (1:3, v/v) over a period of 5 d.

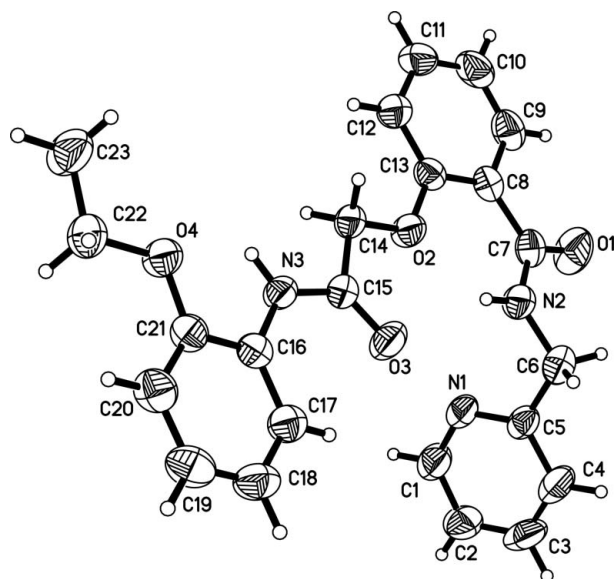


Figure 1
The structure of compound (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.

Crystal data

$C_{23}H_{23}N_3O_4$	$Z = 2$
$M_r = 405.44$	$D_x = 1.344 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 8.7109 (14) \text{ \AA}$	Cell parameters from 1716 reflections
$b = 10.8764 (17) \text{ \AA}$	$\theta = 2.3\text{--}25.5^\circ$
$c = 11.2510 (18) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$\alpha = 106.481 (3)^\circ$	$T = 293 (2) \text{ K}$
$\beta = 90.471 (3)^\circ$	Block, colourless
$\gamma = 100.828 (3)^\circ$	$0.34 \times 0.14 \times 0.08 \text{ mm}$
$V = 1001.8 (3) \text{ \AA}^3$	

Data collection

Siemens SMART 1000 CCD area-detector diffractometer	3863 independent reflections
ω scans	2986 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.015$
$T_{\text{min}} = 0.969$, $T_{\text{max}} = 0.993$	$\theta_{\text{max}} = 26.1^\circ$
5693 measured reflections	$h = -8 \rightarrow 10$
	$k = -13 \rightarrow 13$
	$l = -13 \rightarrow 13$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0469P)^2 + 0.197P]$
$R[F^2 > 2\sigma(F^2)] = 0.052$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.121$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.07$	$\Delta\rho_{\text{max}} = 0.14 \text{ e \AA}^{-3}$
3863 reflections	$\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$
271 parameters	
H-atom parameters constrained	

Table 1

Selected geometric parameters (\AA , $^\circ$).

O1—C7	1.218 (2)	O4—C22	1.431 (2)
O2—C13	1.375 (2)	N2—C7	1.341 (2)
O2—C14	1.4126 (19)	N2—C6	1.443 (2)
O3—C15	1.218 (2)	N3—C15	1.351 (2)
O4—C21	1.363 (2)	N3—C16	1.413 (2)
C7—N2—C6	121.29 (16)	C15—N3—C16	128.42 (15)

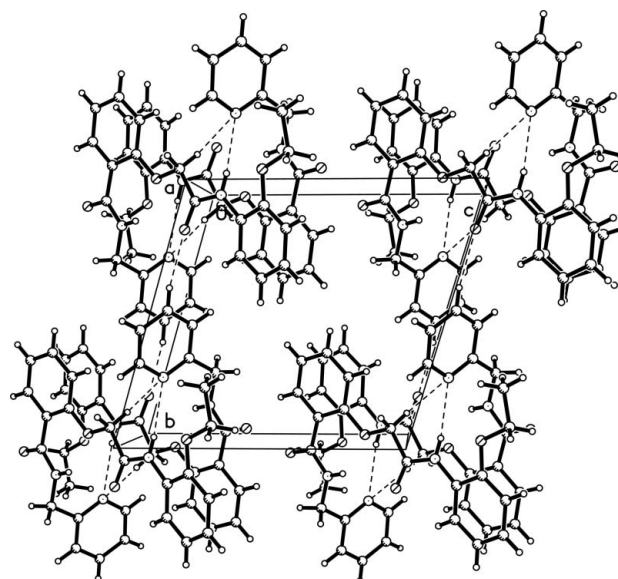


Figure 2
A packing diagram of (I), viewed down the a axis, showing the two-dimensional layers. Dashed lines indicate hydrogen bonds.

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$Cg1$ is the centroid of the pyridine ring (N1/C1—C5).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N2—H2A \cdots O2	0.86	1.98	2.652 (2)	134
N3—H3B \cdots O4	0.86	2.21	2.603 (2)	108
N3—H3B \cdots N1 ⁱ	0.86	2.41	3.224 (2)	158
C3—H3A \cdots O3 ⁱⁱ	0.93	2.36	3.272 (3)	166
C6—H6B \cdots O1	0.97	2.35	2.733 (3)	103
C9—H9A \cdots O1	0.93	2.40	2.739 (3)	101
C14—H14A \cdots N1 ⁱ	0.97	2.54	3.186 (3)	124
C17—H17A \cdots O3	0.93	2.34	2.922 (2)	121
C23—H23B \cdots Cg1 ⁱ	0.96	2.73	3.546 (3)	136

Symmetry codes: (i) $-x + 1, -y, -z$; (ii) $-x + 1, -y - 1, -z$.

All H atoms were located in a difference Fourier map and constrained to ride on their parent atoms, with $C-H = 0.93\text{--}0.97 \text{ \AA}$ and $N-H = 0.86 \text{ \AA}$, and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C or N})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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